SIMPLE AND HIGHLY DIASTEREOSELECTIVE

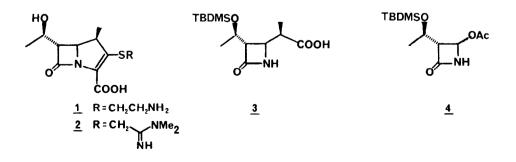
synthesis of a 1 β -methylcarbapenem key intermediate involving divalent tin enolates

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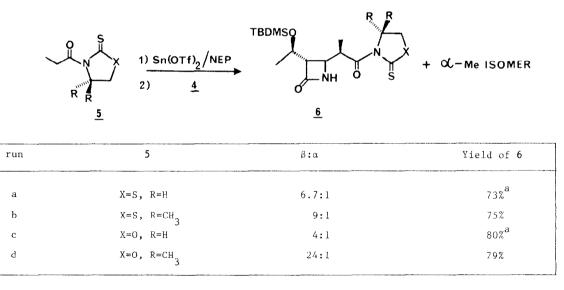
<u>Abstract</u>: A simple and diastereoselective synthesis of 1β -methylcarbapenem key intermediate has been accomplished via a novel C-C bond formation at the C-4 position of 4-acetoxyazetidinone <u>4</u> involving divalent tin enolates of 3-propanoyl thiazolidine and oxazolidine-2-thiones derivatives.

Carbapenems are among the most potent broad spectrum β -lactam antibiotics¹, but many of them are readily metabolized by renal dehydropeptidase-I (DHP-I). On the other hand, Merck researchers found that introduction of a methyl substituent at β C-1 position of the carbapenem nucleus resulted in an exceptional increase of stability, the l β -methylcarbapenems 1 and 2 are good examples^{2a}. However, only a few papers have appeared regarding the stereoselective introduction of the β -methyl substituent². In this communication we wish to report a simple and diastereoselective synthesis of the key intermediate 3 from the readily available azetidinone 4^3 .



Initially, we were especially attracted to Mukaiyama's divalent tin enolates of 3-acyl thiazolidine-2-thiones⁴ because of their **erythro**-selectivity in aldol type reactions and their effectiveness as an active carbonyl group for subsequent transformations. We therefore undertook to test the reaction of the divalent tin enolate of the 3-propanoyl derivative 5a with 4 (Scheme 1, run a). The tin (II) enolate of 5a was prepared according to Mukaiyama's procedure⁴ (5a + N-ethylpiperidine (NEP) and Sn(OTf)₂ in CH₂Cl₂, -20°C + 0°C). Then an acetonitrile solution⁵ of 4 was added and the reaction mixture stirred at 0°C for another 2 h. After work-up, analysis of the crude 6a (HPEC, NNR) indicated a mixture of β and α isomers in a 6.7:1 ratio. After silica gel chromatography the pure diastereomeric mixture was obtained in 73% yield⁶. Treatment of this mixture with 1N sodium hydroxide in THF at 0°C gave the expected acid 3 (β + α) from which the stereochemistry was unambiguously assigned β by comparison with authentic sample^{2a}. Seeking to improve selectivity we next tried the bulkier 4,4-dimethyl substituted derivative 5b⁷. Under similar conditions^{8a}, treatment of 4 with the tin (II) enolate of 5b gave 6b in higher diastereoselectivity (i.e. 9:1). Purification of the latter by silica gel chromatography afforded the pure β isomer 6b in 78% yield as a light yellow solid^{8b}.

SCHEME 1



a) yield includes $\boldsymbol{\alpha}$ isomer.

Encouraged by these results, we decided to investigate the corresponding readily available 1,3-oxazolidine-2-thiones derivatives 5c and 5d as auxiliaries⁹. Accordingly 5c was treated with $Sn(OTf)_2$ and NEP in THF^{10} at -20°C and, after warming to 0°C, 4 was added and the mixture stirred for 30 min. After work-up, analysis of the crude product showed a modest selectivity

of 4:1 for the desired β isomer 6c. We next investigated the corresponding bulkier 4,4-dimethyl substituted derivative 5d. Accordingly, 5d (3.00 mmol) was treated with Sn(OTf)₂ (3.20 mmol) and NEP (3.5 mmol) in dichloromethane (6.8 mL) at 0°C for 30 min. A solution of <u>4</u> (2.0 mmol) in acetonitrile (3.4 mL) was added and the resulting mixture stirred at 0°C for 90 min. Diluted NH₄Cl solution was added and the mixture was stirred vigorously at 0°C for a few min before filtration through Celite. After extraction with EtOAc, analysis of the crude product indicated the formation of the expected intermediate 6d with a remarkable β -diastereoselectivity of 24:1. After silica gel chromatography (CH₂Cl₂-CH₃CN 9:1) pure β isomer 6d (>98.5%) was isolated in 79% yield as a white solid¹¹. Removal of the oxazolidine-2-thione moiety of 6d afforded the desired acid 3 in 89% yield¹².

Thus we have demonstrated that divalent tin enolates of 3-propanoyl thiazolidine and oxazolidine-2-thiones derivatives react with 4 to provide compound 6 with good β -diastereoselectivity particularly when the inexpensive and readily available derivative 5d is used.

Acknowledgements: The authors thank Drs M. Caron and Y.G. Perron and Mr. R. Droghini for their help in preparing this manuscript.

Notes and references

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- 5) Optimum results were obtained when acetonitrile was used as cosolvent.
- 6) Not separable by sílica gel chromatography. However, when the diastereomeric mixutre was treated with ether at 0°C, the insoluble α isomer was easily removed to give almost pure **6a** (98%).
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- 8) (a) Carried out as follows: 1) <u>5b</u> (1.96 mmol), $Sn(OTf)_2$ (2.06 mmol), NEP (2.26 mmol) in CH_2Cl_2 (5 mL) -20°C \rightarrow 0°C 35 min. 2) <u>4</u> (1.30 mmol) in CH_2Cl_2 (1 mL) 0°C 3.5 h. (b) ¹HNMR (200 MHz CDCl_3) & 0.06 (s, 6H, CH_3-Si), 0.86 (s, 9H, CH_3-C-Si), 1.19 (d, 3H, J=7Hz, CH_3-C-H), 1.22 (d, 3H, J=7Hz, CH_3-C-H), 1.57 (s, 3H, CH_3-C-N), 1.63 (s, 3H, CH_3-C-N), 3.09 (dd, 1H, J=2.5, 3.5 Hz, H3), 3.20 (AB, 2H, J=11.3 Hz, CH_2-S), 4.01 (dd, 1H, J=2.5, 3.0 Hz, H4), 4.15 (m, 1H, CH_3-CH-C=0), 4.35 (m, 1H, CH_3CH-O), 5.9 (s, 1H, NH). IR (nujol) 3160,
- 9) Easily prepared from their corresponding commercially available 1,2-aminoalcohols and carbon disulfide in the presence of base. Y. Nagao, T. Kumagai, S. Yamada and E. Fujita, J. Chem. Soc., Perkin Trans. 1, 1985, 2361; Y. Nagao, S. Yamada, T. Kumagai, M. Ochiai and E. Fujita, J. Chem. Soc., Chem. Commun., 1985, 1418 and references cited therein.
- 10) When CH_2Cl_2 was used as solvent, precipitation of the enolate occurred and it reacted sluggishly with 4.
- 11) Spectral data of **6d**. ¹HNMR (200 MHz, CDCl₃) δ 0.06 (s, 6H, CH₃-Si), 0.86 (s, 9H, CH₃-C-Si), 1.20 (d, 3H, J=6.3Hz, CH₃-CH), 1.22 (d, 3H, J=6.9Hz, CH₃-CH), 1.52 (s, 3H, CH₃-C-N), 1.55 (s, 3H, CH₃-C-N), 3.09 (dd, 1H, J=2, 4Hz, H3), 3.95 (dd, 1H, J=2.3, 3.8Hz, H4), 4.16 (s, 2H, CH₂-O), 4.18 (m, 1H, CH₃-CH-C=O), 4.93 (m, 1H, CH₃-CH-O), 5.94 (s, 1H, NH). IR (nujol) 3160, 1760, 1710 and 1335 cm⁻¹.
- 12) To a solution of 6d (0.90 mmol) and 30% H₂O₂ (3 mmol) in THF (6 mL) was slowly added. 1N NaOH (3 mmol) at 20°C. After a few min the mixture was washed with EtOAc, then acidified to pH 2 with conc. HCl. The white solid was collected and dried to give pure <u>4</u> in 89% yield, m.p. 141-142°C.

(Received in USA 22 August 1986)

1760, 1710 and 1300 cm⁻¹